

Ionization and solvation in organic solvents

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Redox reactions in electrolyte solutions depend in the solvation and ionization properties of the solvent and ions. The interplay between the solvation and ionization of lithium ions in various organic solvents is simulated using electron density functional theory (DFT) and simplified models. It is demonstrated, that, overall, DFT is able to predict redox properties of lithium ions in organic solvents. However, there are a few exceptions that require additional treatment if the lithium solvation shell during the ionization. A universal approach of the solvation shell inclusion in the simulation is presented.

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