

Energy and size distribution of $(\text{H}_2\text{O})_n$ and $\text{H}^+(\text{H}_2\text{O})_n$ clusters

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Hydrated ions and water clusters play an essential role in the physics and chemistry of the atmosphere [1]. In this work, the structures and thermodynamic characteristics of $(\text{H}_2\text{O})_n$ and $\text{H}^+(\text{H}_2\text{O})_n$ clusters were studied theoretically. The clusters of size $n \leq 30$ were investigated using the GFN2-XTB method [2].

The results of numerical calculations were used to parameterize the analytical dependence of the cluster energies on their sizes. The data obtained using quantum chemical modeling for $(\text{H}_2\text{O})_n$ and $\text{H}^+(\text{H}_2\text{O})_n$ are well approximated by the analytical functions, which indicates a qualitative agreement between the analytical and numerical models. Comparison of the model parameters with the reference data also indicates a reasonable accuracy of the analytical approximation. Comparison of the calculated size distribution of $\text{H}^+(\text{H}_2\text{O})_n$ clusters with the experimental data from [3] also supports the proposed model.

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- [1] Smirnov B M 2017 *Physics of the global atmosphere. Greenhouse effect, atmospheric electricity, climate evolution (in russian)* (Intellect)
- [2] Bannwarth C, Ehlert S and Grimme S *J. Chem. Theory Comput.*
- [3] Servage K A, Silveira J A, Fort K L and Russell D H 2014 *J. Phys. Chem. Lett.* **5** 1825–1830